Kinetics of microribbon formation in a simplified model of amelogenin biomacromolecules

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We show that the kinetics of microribbon formation of amelogenin molecules is well described by a combination of translational and rotational diffusion of a simplified anisotropic bipolar model consisting of hydrophobic spherical colloid particles and a point charge located on each particle surface. The colloid particles interact via a standard depletion attraction while the point charges interact through a screened Coulomb repulsion. We study the kinetics via a Brownian dynamics simulation of both translational and rotational motions and show that the anisotropy brought in by the charge dramatically affects the kinetic pathway of cluster formation and our simple model captures the main features of the experimental observations.

The self-assembly of colloidal particles and globular 44 tures the main features of the experimental observations 9 proteins with inhomogenous properties into various de- 45 in the early stages. 10 sired structures is an important area of current research, 46 11 in which the inhomogeneity (such as an amphiphilic ₁₂ group and a residual group) plays a crucial role in the 13 kinetic pathway of cluster formation and eventually im-14 pacts on the stabilized crystal structures. Examples in-15 clude Janus particles and DNA-coated colloidal particles 16 that lead to the next generation of building blocks of new 17 materials and have potential applications in fabricating 18 phontonic crystals, targeted drug delivery, and electronic 19 equipment [1-4]. Another important example is amelo-20 genin, the chief hydrophobic protein in enamel matrix 21 with a hydrophilic 25-amino acid C terminus. Amel-22 ogenin is involved in the mineral deposition and is re-23 sponsible for the major structural process during enamel 24 biomineralization. A recent experiment shows that the 25 hydrophilic C-terminus is essential for the self-assembly 26 of amelogenin into microribbons, which may also be rel-27 evant to the elongated and oriented growth of apatite 28 crystals during biomineralization [5, 6]. The experiment 29 also revealed that the microribbon is formed by a hierarchical organization of amelogenin molecules into a chain of nanospheres.

In this article we present a coarse-grained model of amelogenin and explain how the microribbon forms through self-assembly. The amelogenin molecule is hy-35 drophobic with a charged hydrophilic tail. Our model de-36 scribes this in a simplified way, representing the monomer 37 as a spherical molecule, with the charged hydrophilic tail 38 replaced by a single tethered point charge located on the 39 surface of the molecule. Using Brownian dynamics simu-40 lations, we investigate the static and dynamic properties of the self-assembly process. We show that the anisotropy 42 brought by the charge dramatically affects the kinetic

In the experimental studies of amelogenin assembly 47 process, one typically adds salt and a precipitant such 48 as polyethylene glycol (PEG). Although PEG-protein in-49 teractions are quite complicated [7], we model this as 50 a depletion interaction that we take to be given by the 51 Asakura-Oosawa potential plus a repulsive hard-core-like 52 interaction, depending on the center-center distance between spherical particles, $U_c(r_{ij}^c) = U_{AO}(r_{ij}^c) + U_{hc}(r_{ij}^c)$,

$$\frac{U_{AO}(r_{ij}^c)}{kT} = \begin{cases} \phi_p(\frac{r_c}{\xi})^3 \left[\frac{3r_{ij}^c}{2r_c} - \frac{1}{2} \left(\frac{r_{ij}^c}{r_c} \right)^3 - 1 \right], & r_{ij}^c < r_c \\ 0, & r_{ij}^c > r_c \end{cases}$$

55 The cut-off range $r_c \equiv 1 + \xi$, where ξ is the size-ratio 56 between a PEG coil and a colloidal amelogenin particle 57 that controls the range of the depletion interaction, and 58 ϕ_p is the value of the PEG volume fraction that controls 59 the strength of the interaction described by the absolute ovalue of the minimum potential depth $U_m \equiv |U_{min}|$ [8, 9]. 61 The hard core potential is given by

$$\frac{U_{hc}(r_{ij}^c)}{{}_{hT}} = (r_{ij}^c)^{-\alpha}. (2)$$

62 We set $\alpha = 36$, since the values of $\alpha < 36$ have been 63 reported to lead to anomalies when a mimic of the hard-64 core potential is required in the potential [10]. The 65 point charges interact with each other through a screened Coulomb potential,

$$U_p(r_{ij}^p) = \frac{\varepsilon}{r_{ij}^p} \exp\left(-\frac{r_{ij}^p}{\lambda_D}\right),\tag{3}$$

67 in which magnitude controlled by ε and range controlled by Debye screening length λ_D , are varied in the simu-69 lations. We find that the early morphology of the self-43 pathway of cluster formation and our simple model cap- 70 assembly is more sensitive to the size of λ_D rather than

These Coulomb charges exert torques on adjacent molecules and hence produce a rotational motion of the 73 molecules that is included in our Brownian equations of 74 motion, read as:

$$m\ddot{\vec{r}_i} = -\vec{\nabla}(U_i^c + U_i^p) - \Gamma_t \dot{\vec{r}_i} + \vec{W}_i(t), \tag{4}$$

$$I\dot{\vec{\omega}}_i = \vec{\tau}_i - \Gamma_r \vec{\omega}_i + \vec{W}_i'(t), \tag{5}$$

where $m, I, \vec{r_i}, \vec{\omega_i}, \vec{\tau_i}$ are the mass, moment of inertia, position vector, angular velocity, and torque, respectively, of the ith colloidal particle. The mass of the point 79 charge is ignored in this model. Γ_t (Γ_r) is the transla-80 tional (rotational) friction coefficient. \vec{W}_i and \vec{W}'_i are the random forces and torques acting on the ith col-82 loidal particle respectively, which satisfy a fluctuationdissipation relation $\langle \vec{W}_i(t) \cdot \vec{W}_i(t') \rangle = 6kT\Gamma_t \delta_{ij} \delta(t-t'),$ 84 $\langle \vec{W}_i'(t)\cdot\vec{W}_j'(t')\rangle=6kT\Gamma_r\delta_{ij}\delta(t-t')$ [11]. We choose 85 $\Gamma_t=0.5,$ $\Gamma_r=0.167$ and the time step $\triangle t=0.005$ in ₈₆ reduced time units of $\sigma(m/kT)^{1/2}$ with m=1 [12]. The 87 other parameters are chosen according to the range of values of the experimental data displayed in Du et al's 89 paper [6]. In particular, here the results shown are for ₉₀ the values ξ =0.1, $U_m = 6kT$, $\lambda_D = 0.4$ (all length scales are measured in units of monomer diameter and energies ₉₂ are scaled by ε). We consider a small volume fraction of the molecules, typically of the order of f = 0.01 or 0.02. Periodic boundary conditions are enforced to minimize wall effects. All simulations start from a random initial monomer conformation and the results for the kinetics are averaged over several (5-10) runs.

We show typical configurations in the early stage of microribbon formation in Fig. 1. We see that ini-100 tially the monomers form oligomers and the oligomers 101 self-assemble into larger aggregates contained (approxi-102 mately) within nanospheres, then nanospheres associate together into a chain structure, quite similar to the process of amelogenin self-assembly (see Fig. 3 in reference 105

The Coulomb forces cause the majority of the charges 108 to point outwards inside the nanospheres, as shown in 2. This again is reminiscent of the situation 125 111 ize the orientational distribution of point charges within 127 of S(q,t) shown in Fig. 3(a) is given by a slope of - $_{112}$ a nanosphere, we introduce a parameter θ , defined as $_{128}$ 2 on a log-log plot over a reasonable range of q. This $\theta = \cos^{-1}\left(\frac{\vec{u}_p \cdot \vec{u}_c}{|\vec{u}_p||\vec{u}_c|}\right)$, where \vec{u}_p is the position vector of ¹²⁹ value of \vec{D}_f is larger than the typical diffusion-limited¹³⁰ cluster-cluster aggregation (DLCA) value of 1.8 as the 114 a point charge referenced to the center of its host colloid 131 ribbon-like clusters observed here have significant short and \vec{u}_c the position vector of a colloid referenced to the 132 range ordering [13]. For this reason we speculate that 116 center of mass of this nanosphere. Therefore $\theta < 90^{\circ}$ 133 the repulsive Coulomb force causes the surface particles $_{117}$ if the charge points outwards and $\theta > 90^{\circ}$ if it points $_{134}$ to reorganize, which reduces the surface roughness of our 118 inwards. Due to fluctuations and our choice of short re- 125 model, in contrast to the case without the Coulomb in-₁₁₉ pulsion interaction range, the distribution of θ is spread ₁₃₆ teractions (see reference [12]). We find that a peak de-120 around a small nonzero peak position, as shown in Fig. 137 velops in the structure factor whose magnitude increases 121 2, for the configuration in inset. Subsequent aggregation 138 with time. Correspondingly, the peak position decreases 122 forms via necks between nanospheres, leading to larger 139 as a function of wave number with increasing time; these

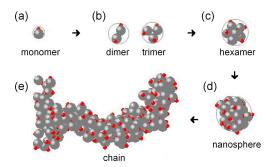


FIG. 1. (color online). (a) The model of amelogenin molecule consists of a spherical colloid particle and a point charge located on its surface that preserves a bipolar nature. (b) and (c) Oligomerization of amelogenin molecules occurs by means of hydrophobic interactions and modification due to Coulomb repulsions. (d) Nanosphere structures are formed through aggregation of monomers and oligomers. (e) Further association of nanospheres results in larger assemblies among which chains are formed.

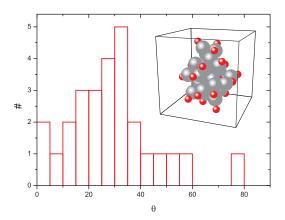


FIG. 2. (color online). Distribution of θ ; θ shows whether the point charge orients outwards ($\theta < 90^{\circ}$) or inwards ($\theta > 90^{\circ}$). Inset: Nanosphere structure, large (grey) spheres indicate amelogenin molecules while small (red) ones for point charges.

We characterize the morphology of the clusters in with the hydrophilic tails of amelogenin. To character- 126 terms of their fractal dimension D_f . The q-dependence 123 structures that eventually form (flexible) microribbons. 140 features are typical of spinodal decomposition [14]. How142 growth considered in this model, the system is controlled 163 in straight simulation from a quench $(U_m = 6kT)$ into the 143 by two characteristic lengths that evolve differently in 164 two phase gas-solid region, we "heat" the system formed ₁₄₄ time and an apparent scaling for the structure factor can ₁₆₅ after 5000 steps from the initial quench $U_m = 6kT$ to ₁₄₅ only be observed over some period of time when these ₁₆₆ $U_m = 4kT$ and run the simulation for another 5000 time 146 two characteristic length scales become comparable to 167 steps. This heating corresponds to the change in PEG 147 each other [15, 16].

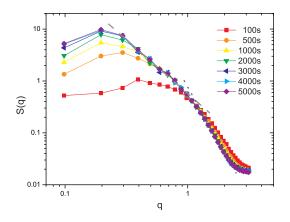


FIG. 3. (color online). Log-log plot of structure factor at several times. The dashed line indicates fractal clusters with $S(q) \propto q^{-D_f}$, where $D_f = 2.0$, while the dotted line indicates the Porod regime $S(q) \propto q^{-(d+1)}$, d=3.

The kinetics of the cluster growth process in this simple model for amelogenin self-assembly is consistent with a cluster-cluster aggregation mechanism, as shown in Fig. 4. The number of clusters decreases inversely with time (i.e. the kinetic exponent z=1) and the radius of gyration increases as a power law with an exponent of n=0.5[17, 18]. The relation between z and n involves the fractal dimension in the following way: $n = z/D_f$. Thus the 158 kinetic exponents are consistent with a fractal dimension 159 of $D_f = 2$ as well.

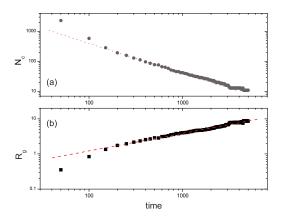


FIG. 4. (color online). Plot of (a) number of clusters, N_c and (b) radius of gyration, R_q as a function of time in a log-log 182 λ_2^2 , and λ_3^2 in descending order [19]. We calculate the raplot. The dotted line in (a) has a slope of -1; the dashed line 183 tios among these three principal values, which characterin (b) has a slope of 0.50.

141 ever, we note that for deep quenches that lead to cluster 162 In order to check the stability of the clusters produced 168 concentration as it is done in the experiment. A typi-169 cal run is shown in Fig. 5. As shown in the figure, we 170 find that the microribbons are quite robust under such a 171 change of conditions, as found experimentally for a wide 172 range of conditions (including a change in the PEG con-173 centration). We have also checked the structure factor behavior for the heating process and found that the struc-175 ture factor curve has the same exponent of -2 as at the 176 cooler temperature (see Fig. 3). This is consistent with 177 the robustness of the major cluster morphology.

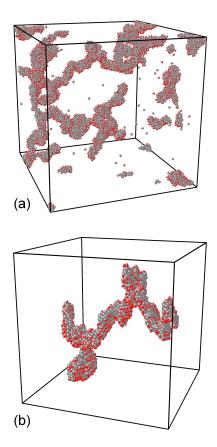


FIG. 5. (color online). (a) Morphology of entire system ($\xi =$ 0.1, f = 0.02) from another 5000 time steps shallow quench of $U_m = 4kT$ after first quench into $U_m = 6kT$, 5000 time steps. (periodic boundary condition are enforced). (b) One of the clusters inside (a).

We characterize these microribbons by computing three eigenvalues of the gyration tensor, denoted as λ_1^2 , 184 ize the relative colloid (or charge) particle distribution. A 185 typical result is given in Table 1, for the colloid particles

TABLE I. Typical ratios of eigenvalues of gyration tensor

	λ_1^2/λ_3^2	λ_2^2/λ_3^2
Colloid particles	9.57	9.03
Point charges	9.25	8.74

and point charges separately, based on the configuration in Fig. 5(b). These ratios of eigenvalues demonstrate the anisotropy of these clusters, suggesting that the clusters ter is ribbon-like. The difference of the ratios between colloidal particles and point charges may indicate that under electrical repulsion, the rotation of monomers in side the cluster leads to the small reduction of anisotropy in the point charges' spatial distribution.

From the gyration tensor we can determine the typical 195 length scales of these clusters by taking square roots of 239 the eigenvalues. To illustrate this, we use the colloid con-198 figuration shown in Fig. 5(b), which gives 18.27, 17.75, 241 and 5.91 for the three length scales. Since the chain struc-200 ture is the association of several nanospheres (as shown 201 in Fig. 1), the typical diameter of the nanosphere in 202 this configuration is at the most 5.91. We can also use this estimate of the diameter to obtain an estimate of the minimum wavenumber that corresponds in Fig. 3 248 [10] to the crossover to the Porod regime (corresponding to ²⁴⁹ compact clusters on a local scale). This yields a value of $q_{min} \simeq 2\pi/5.91 = 1.06$, which is consistent with the behavior shown in Fig. 3(b). Since the hydrodynamic $\frac{252}{253}$ 209 radius R_H of the amelogenin macromolecule is about 2.2 $_{254}$ 210 nm, the R_H of the nanosphere for the configuration in 255 [13] $_{211}$ Fig. 5(b) is $2.2 \times 5.91 = 13.00$ nm; this lies in the range $_{256}$ $_{212}$ of values of the radii of typical nanospheres, $10 \sim 25$ nm, $_{257}$ found in the experiment (cf. Fig. 3, reference [6]).

In summary, we have developed an anisotropic, bipolar model for the hierarchical self-assembly of amelogenin
lar molecules and have carried out Brownian dynamics simlar ulations of the self-assembly process. Simulations show
lar a hierarchical self-assembly process where the molecules
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²²⁴ are stable against microribbon formation. Thus a con-²²⁵ trol over the morphology of the clusters can be achieved. ²²⁶ Such a study is currently underway.

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- [1] Qian Chen, Sung Chul Bae, and Steve Granick, Nature **469**, 381, (2011).
- [2] Flavio Romano and Francesco Sciortino, Nature material **10**, 171, (2011).
- [3] L. Hong, A. Cacciuto, E. Luijten, and S. Granick, Langmuir 24, 621-625 (2008).
- [4] Chad A. Mirkin, MRS Bulletin 35, 532-539 (2010).
- [5] S. Gajjeraman, K. Narayanan, J. Hao, C. Qin and A. George, J. Biological chemistry 282, 1193 (2007).
- [6] C. Du, G. Falini, S. Fermani, C. Abbott, J. Moradian-Oldak, Science 307, 1450 (2005).
- [7] J. Bloustine, T. Virmani, G. M. Thurston, and S. Fraden, Phys. Rev. Lett. 96, 087803 (2006).
- [8] S. Asakura and F. Oosawa, J. Chem. Phys. 22, 1255 (1954).
- [9] A. Vrij, Pure Appl. Chem. 48, 471 (1976).
- [10] K. G. Soga, J. R. Melrose, and R. C. Ball, J. Chem. Phys. 108, 6026 (1998); K. G. Soga, J. R. Melrose, and R. C. Ball, J. Chem. Phys. 110, 2280 (1999).
- ¹ [11] W. F. Van Gunsteren, and H. J. C. Berendsen, Mol. Phys. 45, 637 (1982).
- [12] Wei Li, J. D. Gunton, Siddique J. Khan, J. K. Schoelz,
 and A. Chakrabarti, J. Chem. Phys. 134, 024902 (2011).
- [13] G. Porod, Zeit. Kolloid, 124, 53 (1951). See also, O. Glatter and O. Kratky. Small-angle X-ray Scattering, (Academic Press, London 1982).
- [14] J. D. Gunton, M. San Miguel, and P. S. Sahni, in Phase Transitions and Critical Phenomena, edited by C. Domb and J. L. Lebowitz (Academic, London, 1983), Vol. 8.
- ¹ [15] J. J. Cerda, T. Sintes, C. M. Sorensen, and A. Chakrabarti, Phys. Rev. E **70**, 051405 (2004).
- [16] H. Huang, C. Oh, and, C.M. Sorensen, Phys. Rev. E 57, 875 (1998).
- [17] S. K. Friedlander. Smoke, Dust, and Haze: Fundamentals of Aerosol Behavior (Wiley, New York, 1977).
- [18] S. J. Khan, C.M. Sorensen, and A. Chakrabarti, J. Chem.
 Phys. 131, 194908 (2009).
- [19] Doros N. Theodorou and Ulrich W. Suter, Macromolecules, 18, 1206 (1985).